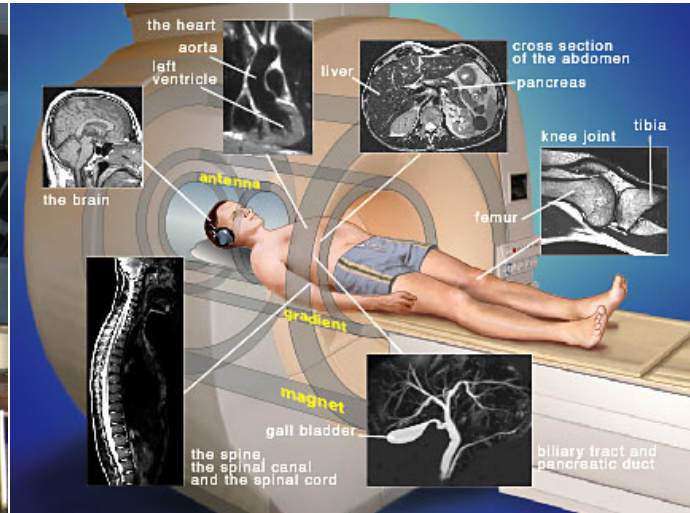


# Physics 2049, General Physics B

Prof. Stephen Hill, Course Leader

*An Introduction to Electricity, Magnetism and Optics*



**Prof. H-K. Ng**



**Prof. O. Vafek**



**Prof. S. Tabor**



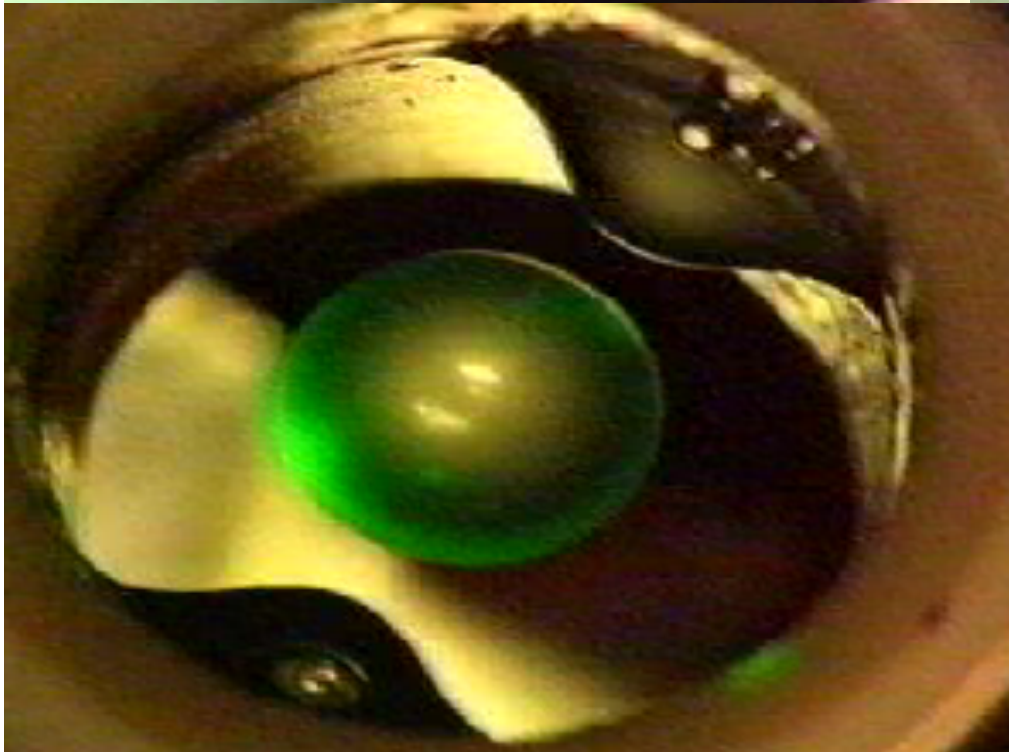
# Physics 2049C: This Week and Next

- No labs this week or next; they start in week 3.
- First LON-CAPA assignment due Thursday!!
  - You will prepare for this in tomorrow's recitation
  - Very strict deadline of 11:59:59 pm on due date
  - See syllabus and LON-CAPA for subsequent deadlines
- First Mini-Exam next Thursday (Sep. 8<sup>th</sup>)
- Today:
  - Intro to Electricity and Magnetism
  - Start Ch. 20 - Electric charge and force
- Thursday:
  - Cont... electric field and Coulomb's law

# PHY2048 - What causes acceleration?

## Linguistic arguments:

- Some sort of interaction - loosely speaking, a push or a pull on an object.
- We call this a force, which can be said to act on a body.
- Examples of forces:
  - Normal or "contact force"
  - Gravitational force
  - Electromagnetic force
  - Weak and strong nuclear forces

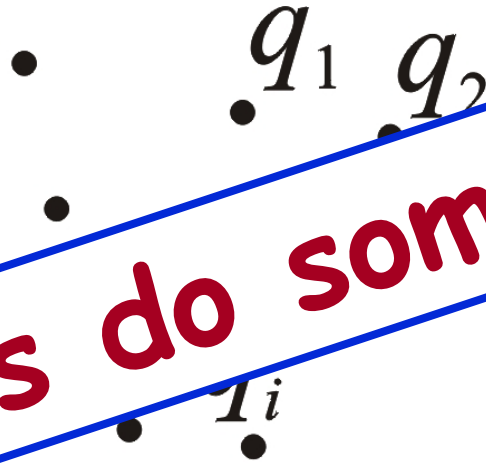


**We'd better be  
more scientific  
about this...**

The main subject of this course

## Electrostatics

Source  
charges



Test charge

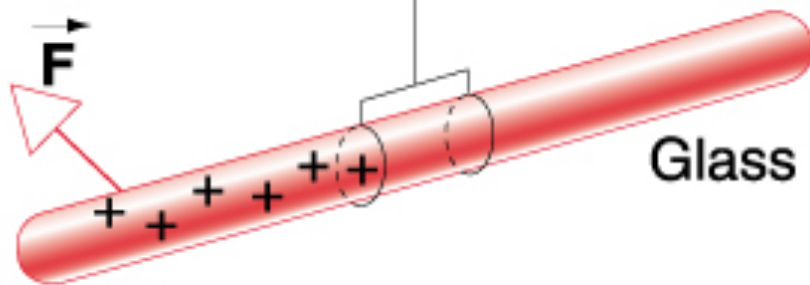


Let's do some experiments

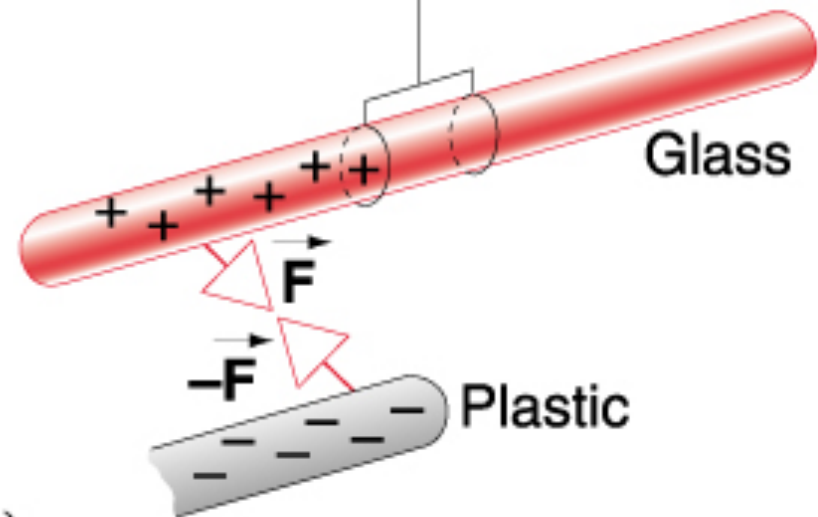
- What is the force on  $q_0$ ?
- How much energy would I have to supply to move  $q_0$  a distance  $r$ ?

## Charges exert forces on each other

- *Charges of the same sign repel one another, charges of opposite sign attract one another.*



(a)



(b)

# What is charge?

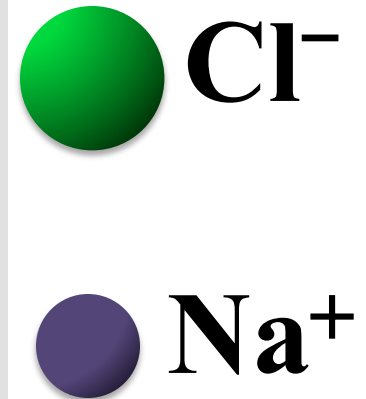
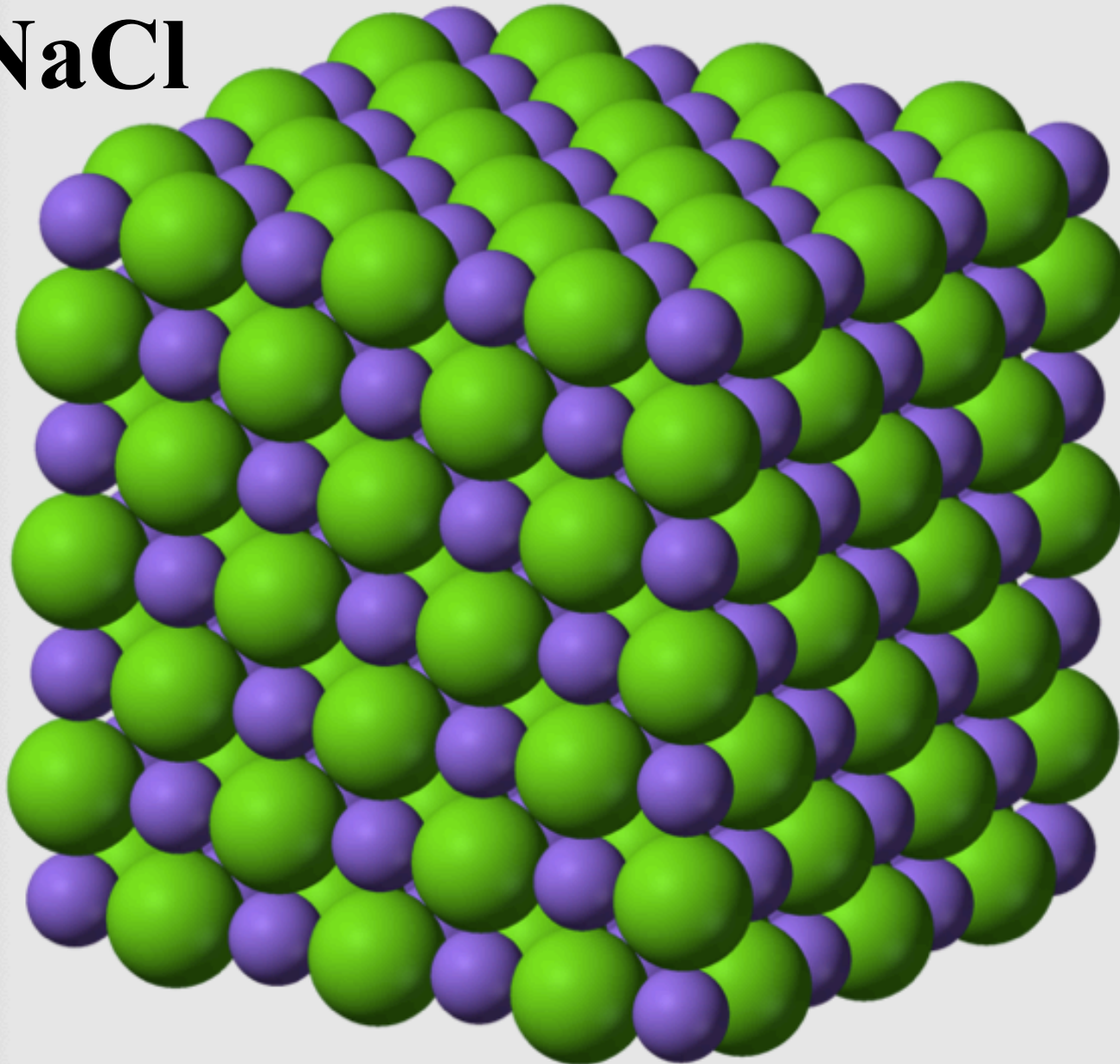
- Charge is measured in Coulombs (C)
  - Fundamental unit.
  - Definition based on forces between current carrying wires (current = Ampères, or C/s), i.e. chapter 26.
- Charge is discrete
  - Thompson discovered the electron in 1896. He found that charge was carried by elementary particles with the same charge to mass ratio.
  - The elementary charge of the electron was not measured until 1909 (Millikan).
  - Both experiments earned Nobel prizes.

Charge on an electron:	$e = 1.6 \times 10^{-19}$ Coulombs
1 Coulomb of charge:	$6.24 \times 10^{18}$ electrons
1 Ampère (= 1 C/s)	$6.24 \times 10^{18}$ electrons/second

Charge is discrete:  $q = ne$        $n = \pm 1, \pm 2, \pm 3, \dots$

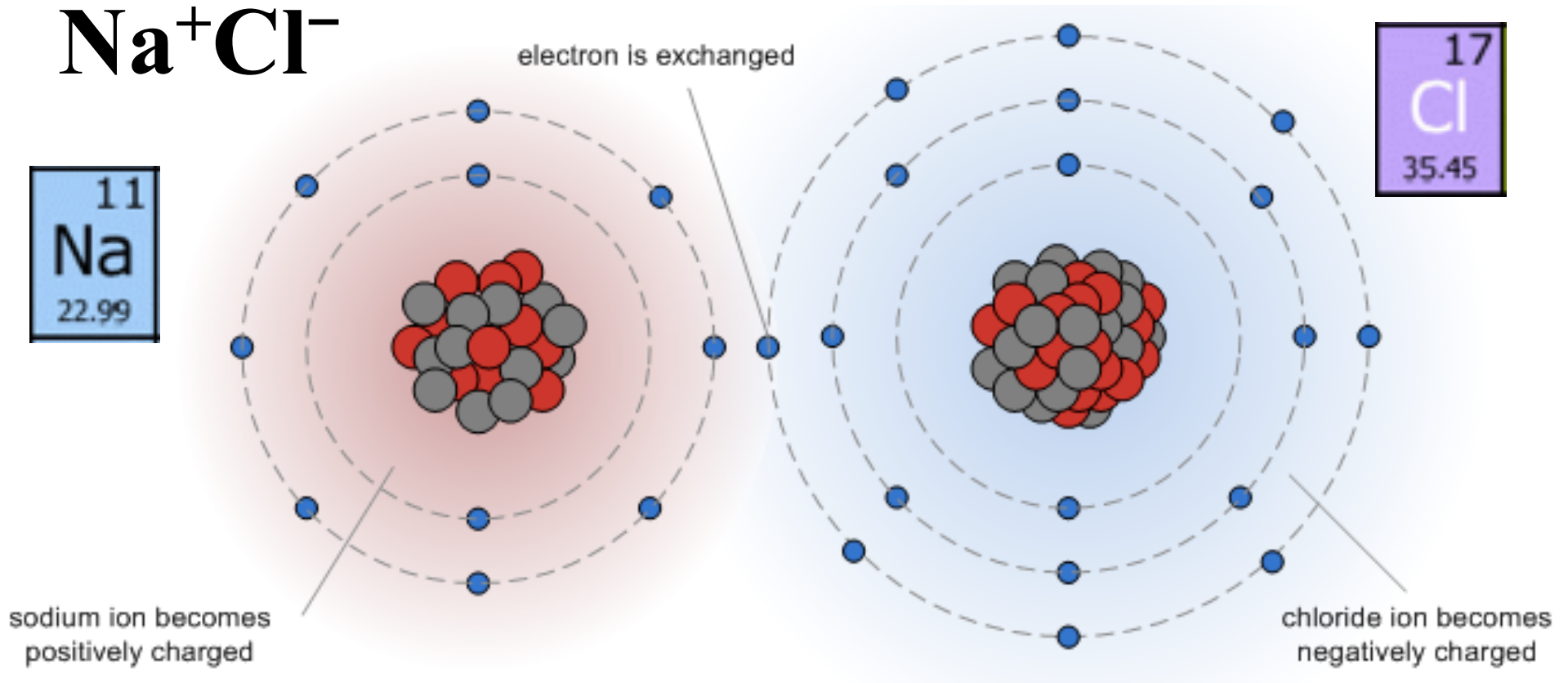
# Matter - solids, liquids and gases

**NaCl**



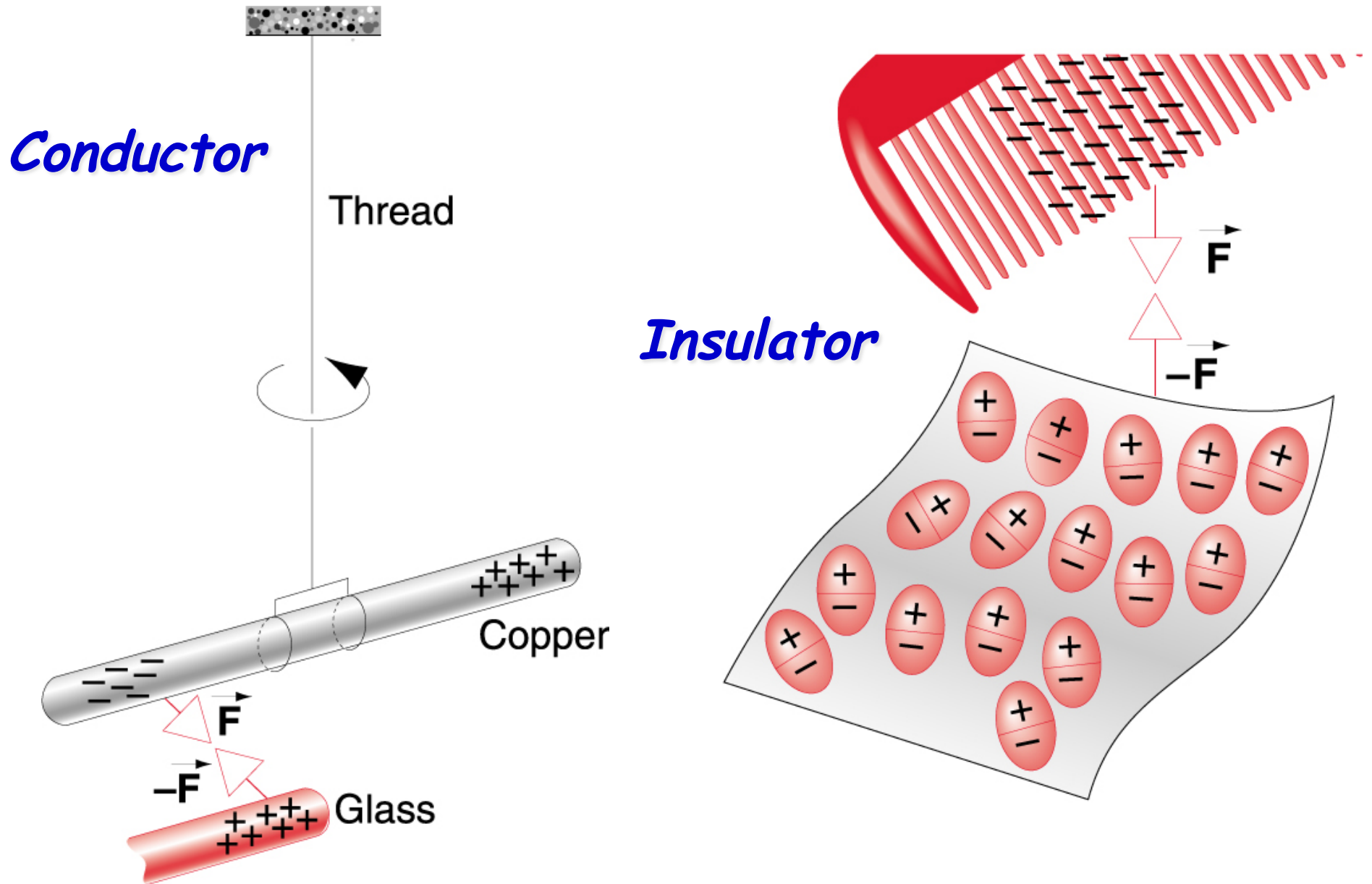


# Matter - solids, liquids and gases

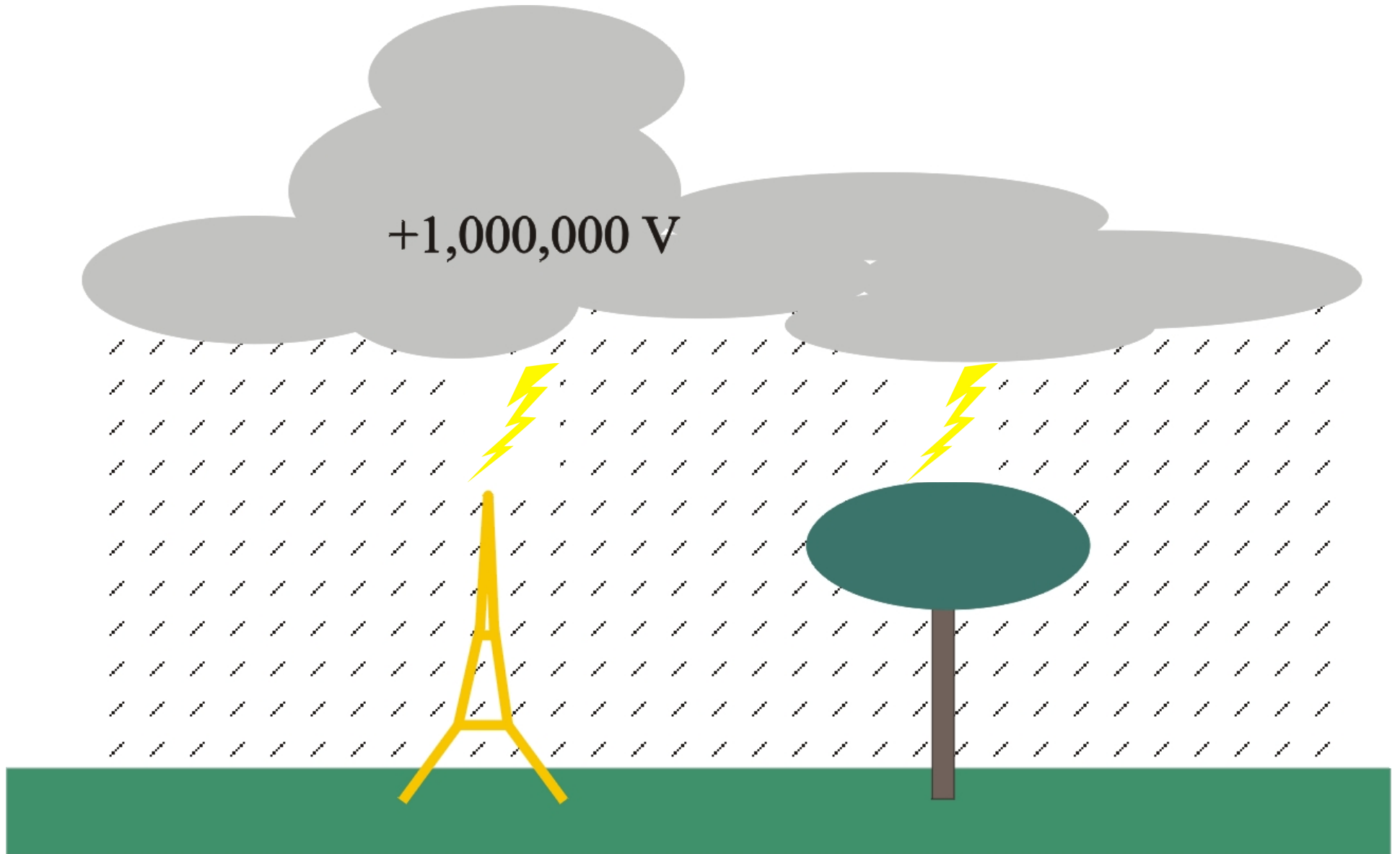


- Proton: charge =  $+1.6 \times 10^{-19}$  Coulombs
- Neutron: charge = 0 (neutral)
- Electron: charge =  $-1.6 \times 10^{-19}$  Coulombs

# Static electricity through charging



# Electrical discharge



# History Lesson

- 600BC** **Greek philosophers**  
First references to magnetism and electric charge
- 1175-1600** **Alexander Neckem, Petrus Peregrinus, William Gilbert**  
References to and explanation for the compass
- 1747** **Benjamin Franklin (and William Watson)**  
Discovers that there are two kinds of charge
- 1780s** **Charles Augustine de Coulomb**  
Discovers law of forces between charges – birth of electrostatics
- 1825** **André-Marie Ampère**  
Discovers law of forces between currents – birth of magnetostatics
- 1720** **Hans Christian Oersted**  
Discovers that electric currents influence compass needles
- 1831** **Michael Faraday**  
Discovers law of electromagnetic induction – birth of electrodynamics
- 1873** **James Clerk Maxwell**  
Publishes *A Treatise on Electricity and Magnetism*
- 1887** **Henrich Hertz**  
Confirms that light is an electromagnetic wave
- 1905** **Albert Einstein\***  
Formulates special theory of relativity
- 1909** **Robert Millikan\***  
Measurement of elementary unit of charge

**\*Nobel  
prize**

# PHY3101 - next semester

At the end of the 19th century, A. A. Michelson (very famous physicist) stated that *"all of the grand underlying physical principles had been firmly established."*

Then came two revolutions:

- Relativity

*concepts of space and time change at large relative velocities*

- Quantum mechanics

*concept of matter changes on small length scales*

- Classical laws of mechanics break down in these limits, and much remains to be discovered

# Coulomb's Law

## *Coulomb's torsional balance*

1785



$$F \propto \frac{|q_1||q_2|}{r^2}, \quad \text{or} \quad F = k \frac{|q_1||q_2|}{r^2}$$

$$k = \frac{1}{4\pi\epsilon_0} = 8.99 \times 10^9 \text{ N} \cdot \text{m}^2 / \text{C}^2$$

$$\epsilon_0 = 8.85418781762 \times 10^{-12} \text{ C}^2/\text{N} \cdot \text{m}^2$$

$$\epsilon_0 = \frac{1}{\left(4\pi \times 10^{-7} \text{ N} \cdot \text{s}^2 / \text{C}^2\right) \times c^2}$$

$$\begin{aligned} c &= \text{speed of light in vacuum} \\ &= 299792458 \text{ m/s} \end{aligned}$$